## Determining n(T) for Extrinsic Materials by Dr. Colton, Physics 581 (last updated: Fall 2020)

### **Intrinsic reminder**

As a reminder, for intrinsic materials we were able to deduce the number of electrons in the conduction band as a function of temperature, n(T) for a given material (with a given band gap), by writing these equations for n and p, and using the equation n = p to connect the two.

$$n = 2 \left(\frac{m_e kT}{2\pi\hbar^2}\right)^{3/2} e^{-(E_c - \mu)/kT}$$
$$p = 2 \left(\frac{m_h kT}{2\pi\hbar^2}\right)^{3/2} e^{-(\mu - E_v)/kT}$$
$$n = p \quad \text{(intrinsic condition)}$$

Those were the three governing equations for the intrinsic case.  $E_c$  and  $E_v$  are the conduction and valence band energies, respectively.

None of the equations individually gives enough information to solve for n(T), because we don't know how  $\mu$  varies with temperature (at least, we didn't at first). However, multiplying the first two equations together allows us to determine n(T) by recognizing that  $np = n^2$  is true, from the intrinsic condition. As an added bonus, we could also determine  $\mu(T)$  by equating the first two equations and solving for  $\mu$ .

Alternatively, we could have first determined the equation for  $\mu(T)$ , and then plugged that back into the *n* equation. That would have led to the same result for n(T) that we obtained by multiplying *n* and *p* together. We'll take that second approach for extrinsic (doped) materials: first find  $\mu$ , then plug that value of  $\mu$  into the equation for *n*, to find *n*.

#### Governing equations for extrinsic materials

Let's now consider an n-type doped material; hopefully the extrapolation to a p-type material will be clear should you ever need to do that. For an n-type material, the above equations for n and p are still the same,<sup>1</sup>

$$n = 2 \left(\frac{m_e kT}{2\pi\hbar^2}\right)^{3/2} e^{-(E_c - \mu)/kT}$$
$$p = 2 \left(\frac{m_h kT}{2\pi\hbar^2}\right)^{3/2} e^{-(\mu - E_v)/kT}$$

but the equation connecting the two is now different:

$$n = p + N_d^+$$
 (extrinsic condition)

<sup>&</sup>lt;sup>1</sup> This is actually an approximation, since technically in deriving the earlier equations for n and p we had to make a small temperature approximation which may or may not exactly apply now. But we'll ignore that, since everyone else does, and the resulting equations work well.

Here  $N_d^+$  is the concentration of ionized donors, i.e. the number of donors per volume whose electrons have been promoted to the conduction band. A formula for  $N_d^+$  can be obtained via a thermodynamic treatment of ionized donors, as the following:

$$N_{d}^{+} = \frac{N_{d}}{1 + 2e^{(\mu - E_{d})/kT}}$$

where  $N_d$  is the total concentration of donors and  $E_d$  is the absolute energy of the donor level (i.e. not the donor ionization energy, which is the difference between  $E_c$  and  $E_d$ ).

Those are the four governing equations for the extrinsic case.

#### Solving the governing equations for the extrinsic case

We can now put the four equations together by plugging Eqs. 1, 2, and 4 into Eq. 3 like this:

 $n = p + N_d^+$  (extrinsic condition)

$$2\left(\frac{m_e kT}{2\pi\hbar^2}\right)^{3/2} e^{-(E_c - \mu)/kT} = 2\left(\frac{m_h kT}{2\pi\hbar^2}\right)^{3/2} e^{-(\mu - E_v)/kT} + \frac{N_d}{1 + 2e^{(\mu - E_d)/kT}}$$

For a situation where we are given the material and the amount of doping,  $\mu$  and T are the only unknowns in that equation. The equation can therefore in theory then be solved for  $\mu$  as a function of T, and  $\mu(T)$  can then be plugged into the first governing equation to give us n(T).

However, unlike the intrinsic case, that equation is not analytically solvable for  $\mu(T)$ . Therefore we must solve it numerically: pick a temperature, solve numerically for  $\mu$  at that temperature, then plug that  $\mu$  into the first governing equation to get n for that temperature. Let's do that in an example.

# Example: Silicon Doped With $10^{21}$ Donors/m<sup>3</sup>, T = 77 and 300 K

For silicon at room temperature, the quantity  $2\left(\frac{m_e kT}{2\pi\hbar^2}\right)^{3/2}$  equals 2.86 × 10<sup>25</sup> m<sup>-3</sup> and the quantity  $2\left(\frac{m_h kT}{2\pi\hbar^2}\right)^{3/2}$  equals 2.50434 × 10<sup>25</sup> m<sup>-3</sup>. (Stokes calls these values  $N_c$  and  $N_v$ , respectively.) To simplify typing things into Mathematica, I'm therefore going to rewrite the first two governing equations like this:

$$n = 2.86 \times 10^{25} \left(\frac{T}{300}\right)^{3/2} e^{-(E_c - \mu)/kT}$$
$$p = 2.50434 \times 10^{25} \left(\frac{T}{300}\right)^{3/2} e^{-(\mu - E_v)/kT}$$

where the temperatures now must explicitly be given in kelvins and the concentrations in m<sup>3</sup>.

I'm also going to set the valence band energy to zero, as is often done, so  $E_v = 0$ ,  $E_c =$  band gap, and  $E_d = E_g - 0.045$  eV (taking the donor ionization energy for silicon to be 45 meV).

With all of that, I need to then numerically solve this equation for  $\mu$ , then plug that  $\mu$  back into the first governing equation to get *n*. I'll now turn to Mathematica to first solve for 77 K and then for 300 K. Pay attention to the comments in the code.

## Setting things up

```
In[1]:= k = 1.381*^-23;
e = 1.602*^-19;
Eg = 1.124 e; (* band gap of Si in eV, multiplying by e to convert to J *)
Ev = 0; (* picking my zero of energy to be at VB *)
Ec = Ev + Eg;
Ed = Ec - 0.045 e; (* the absolute position of donor level, not relative *)
Nd = 1*^21;
```

(\* I'm going to multiply mu by charge of electron in these next equations so mu in the plot will have units of eV. I will also divide all concentrations by 10^21 for now so the "FindRoot" command doesn't have to work with huge exponents. \*) n[mu\_, T\_] := 2.86\*^25 (T / 300) ^ (3 / 2) Exp[(mu e - Ec) / (k T)] / 1\*^21 p[mu\_, T\_] := 2.50434\*^25 (T / 300) ^ (3 / 2) Exp[(Ev - mu e) / (k T)] / 1\*^21 Ndplus[mu\_, T\_] := Nd / (1 + 2 Exp[(mu e - Ed) / (k T)]) / 1\*^21(\* mu in eV, n in 10^21 \*)

(\* The extrinsic condition is "n = p + Ndplus". I'm going to plot both sides of that equation as a function of mu then find the mu where they intersect. Then I'll use that mu in the n equation to get the answer for a desired temperature. \*)

```
Solving when T = 77K
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```
In[10]:= T = 77;
```





```
In[12]:= (* \text{ Looks like answer must be around mu = 1.05 *})
FindRoot[n[mu, T] == p[mu, T] + Ndplus[mu, T], {mu, 1.05}]
Out[12]:= {mu \rightarrow 1.06744}
First step: \mu(T = 77K) = 1.06744 \text{ eV}
In[13]:= foundmu = mu /. %; (* This is needed because the FindRoot command provides its answer in a
weird format *)
n[foundmu, T] * 1*^21 (* the answer for T=77! *)
Out[13]:= 7.40579 × 10<sup>20</sup>
Second step: n(T = 77K) = 7.40579 × 10^{20} \text{ m}^{-3}
```

## Solving when T = 300K

```
\ln[14]= (* I'll do everything in one cell now. I'll try to use the same starting point for my in
       the FindRoot command that I did for 77 K, namely 1.05. It could fail... but fortunately it doesn't. *)
      T = 300;
      Plot[{n[mu, T], p[mu, T] + Ndplus[mu, T]}, {mu, 0, 1.2}]
      FindRoot[n[mu, T] == p[mu, T] + Ndplus[mu, T], {mu, 1.05}]
      foundmu = mu /.%;
      n[foundmu, T] * 1*^21(* the answer for T=300! *)
      80
      60
Out[15]= 40
                                                              the point where n = p + N_d^+,
      20
                                                              for this new temperature
                                                             1.2
                 0.2
                         0.4
                                                    1.0
                                  0.6
                                           0.8
Out[16]= \{mu \rightarrow 0.858621\}
                                                                          First step: \mu(T = 300K) = 0.858621 eV
                                                              Second step: n(T = 300K) = 9.99602 \times 10^{20} \text{ m}^{-3}
Out[18]= 9.99602 \times 10^{20}
```

To reiterate: you pick a temperature, solve numerically for  $\mu$  for that temperature, then plug that  $\mu$  into the equation for n, and you have the value of n for that temperature.

## Silicon Doped With 10<sup>21</sup> Donors/m<sup>3</sup>, arbitrary temperatures

That same procedure can be employed to determine  $\mu$  for any arbitrary temperature and make a plot of  $\mu(T)$ , and to determine *n* for an arbitrary temperature and make a plot of n(T). I'll let Mathematica do the "pick a temperature" part automatically as it creates these plots for me.

## Solving for arbitrary temperatures

In[19]:= Off[General::munfl] (\* This suppresses a warning message that will otherwise show up \*)
 (\* I'm going to skip making the plot of the two sides of the extrinsic equation for the
 desired temperature and will skip straight to the FindRoot command \*)
 actualmu[7\_] := mu /. FindRoot[n[mu, 7] == p[mu, 7] + Ndplus[mu, 7], {mu, 1.05}];
 Plot[actualmu[T], {T, 5, 1000}, PlotRange → {0, 1.2}]

